

# Machine Learning of Quasiparticle Energies in Molecules and Clusters

Onur Çaylak and Björn Baumeier\*

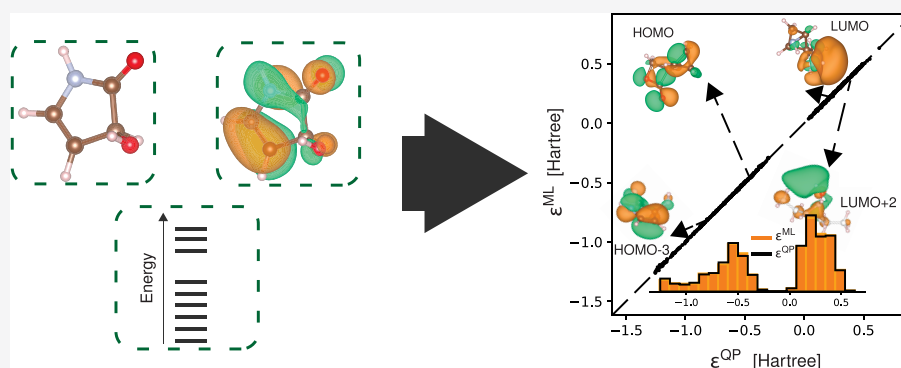
Cite This: *J. Chem. Theory Comput.* 2021, 17, 4891–4900

Read Online

ACCESS |

Metrics & More

Article Recommendations



**ABSTRACT:** We present a  $\Delta$ -machine learning approach for the prediction of GW quasiparticle energies ( $\Delta$ MLQP) and photoelectron spectra of molecules and clusters, using orbital-sensitive representations (OSRs) based on molecular Cartesian coordinates in kernel ridge regression-based supervised learning. Coulomb matrix, bag-of-bond, and bond-angle-torsion representations are made orbital-sensitive by augmenting them with atom-centered orbital charges and Kohn–Sham orbital energies, both of which are readily available from baseline calculations at the level of density functional theory (DFT). We first illustrate the effects of different constructions of the OSRs on the prediction of frontier orbital energies of 22k molecules of the QM8 data set and show that it is possible to predict the full photoelectron spectrum of molecules within the data set using a single model with a mean absolute error below 0.1 eV. We further demonstrate that the OSR-based  $\Delta$ MLQP captures the effects of intra- and intermolecular conformations in application to water monomers and dimers. Finally, we show that the approach can be embedded in multiscale simulation workflows, by studying the solvatochromic shifts of quasiparticle and electron–hole excitation energies of solvated acetone in a setup combining molecular dynamics, DFT, the GW approximation, and the Bethe–Salpeter equation. Our findings suggest that the  $\Delta$ MLQP model allows us to predict quasiparticle energies and photoelectron spectra of molecules and clusters with GW accuracy at DFT cost.

## 1. INTRODUCTION

Fundamental insights gained by computational analysis of electronically excited states of molecular systems can help improve the design of molecular materials and play therefore a vital role in materials science. However, obtaining quantitative predictions is challenging, as traditional methods either come with insufficient accuracy, for example, due to the lack of correlation in interpreting orbital energies of Hartree–Fock, or at the price of high computational costs, as for coupled cluster methods, quantum Monte Carlo, or Green’s function approaches. Hence, the incorporation of quantum machine learning (QML) has been gaining great traction over recent years. QML-based surrogate property models have become a popular alternative approach for their fast, reliable, and accurate predictions of molecular and material properties.<sup>1–6,8–16</sup>

The main advantage of ML models is that they allow predictions of molecular properties with improved efficiency at a lower computational cost compared to traditional quantum chemistry approaches. Method development in the field of QML is progressing rapidly, and it is increasingly influencing traditional methods.<sup>6,17–21</sup> Developments in molecular representations and QML models have paved the way for predicting energetic, electronic, and thermodynamic properties, such as atomization energies, dipole moments, polarizabilities, and harmonic frequencies.<sup>22–24</sup>

Received: May 25, 2021

Published: July 27, 2021



QML of excited states of molecules has remained difficult in comparison. Recent work<sup>21,25,26</sup> has achieved promising results for predictions of single frontier orbital (highest or lowest molecular orbital) energies. However, some applications, such as the evaluation of direct or inverse photoelectron spectra, require predictions for a wider range of orbitals simultaneously, with sensitivity to conformational details of the actual molecules and/or a complex embedding environment. Circumventing the need to build separate models for each state of interest and the associated difficulties in finding unique characterizations of multiple orbitals across a wide range of chemical space can be achieved, for instance, using neural networks that learn the required information implicitly from orbital-free representations<sup>5,7</sup> or by capturing both structural and orbital details in the representation.

In this paper, we show a way of augmenting existing molecular Cartesian coordinate-based representations with orbital-specific information from density functional theory (DFT), which allows us to individually predict quasiparticle energies at the level of many-body Green's function theory in the GW approximation and to calculate full photoelectron spectra of molecules and clusters with a single-target kernel ridge regression (KRR)  $\Delta$ -machine learning model ( $\Delta$ MLQP). We adopt the  $\Delta$ -ML approach<sup>23</sup> as its concept of learning the corrections to a certain baseline property matches directly the way in which quasiparticle corrections are obtained perturbatively to the Kohn–Sham orbital energies, that is,  $\varepsilon_i^{\text{QP}} = \varepsilon_i^{\text{KS}} + \Delta\varepsilon_i^{\text{GW}}$ . Within  $\Delta$ MLQP, we consider specifically the orbital-sensitive augmentation of Coulomb matrix (CM), bag-of-bond (BoB), and bond-angle-torsion (BAT) representations by a combination of atom-centered orbital charges and Kohn–Sham orbital energies, which are all easily accessible from standard DFT ground-state calculations.<sup>2,9,24,27</sup> We illustrate the effect of different methodological choices, such as the determination of the orbital charges as Mulliken populations<sup>28</sup> or from a Gaussian-distributed multipole analysis<sup>29</sup> (GDMA), on the prediction of the full quasiparticle spectra of molecules in the QM8 data set. To scrutinize that orbital-sensitive representations (OSRs) with multiple orbitals are also capable of resolving the effects of intra- and intermolecular conformations, we also study the photoelectron spectra of water monomers and dimers, taken from the H2O-13 data set.<sup>30</sup> Finally, we study the use of the  $\Delta$ MLQP-predicted, conformational-sensitive quasiparticle energies embedded in calculations of electron–hole excitations at the level of the Bethe–Salpeter equation (BSE) for the prototypical example of acetone in aqueous solution and compare the solvatochromic shifts obtained from BSE@GW and BSE@ML.

This paper is organized as follows: In Section 2, we briefly summarize the background of many-body Green's function theory in the GW approximation with the BSE of KRR models and  $\Delta$ -machine learning. Section 3 introduces the  $\Delta$ MLQP approach and showcases results in application to predictions of quasiparticle energies of molecules from the QM8 data set, water monomers and dimers, and acetone in aqueous solution. A brief summary concludes the paper.

## 2. METHODOLOGICAL BACKGROUND

### 2.1. Quasiparticle and Electron–Hole Excitations with GW-BSE.

For a given closed-shell ground-state system with  $N$  electrons, one can within DFT get the Kohn–Sham energies<sup>31</sup> by solving

$$\hat{H}^{\text{KS}}|\varphi_i^{\text{KS}}\rangle = [\hat{H}_0 + \hat{V}_{\text{xc}}]|\varphi_i^{\text{KS}}\rangle = \varepsilon_i^{\text{KS}}|\varphi_i^{\text{KS}}\rangle \quad (1)$$

with  $\hat{H}_0 = \hat{T}_0 + \hat{V}_{\text{ext}} + \hat{V}_{\text{H}}$ , where  $\hat{T}_0$  stands for the kinetic energy,  $\hat{V}_{\text{ext}}$  is the external potential,  $\hat{V}_{\text{H}}$  is the Hartree potential, and  $\hat{V}_{\text{xc}}$  is the exchange–correlation potential.

The addition ( $N \rightarrow N + 1$ ) or removal ( $N \rightarrow N - 1$ ) of a single electron to/from the system can be seen as the excitation of a quasiparticle. Quasiparticle energies are essentially the poles of interacting one-electron Green's function and obey Dyson's equation<sup>32–35</sup>

$$[\hat{H}_0 + \hat{\Sigma}(\varepsilon_i^{\text{QP}})]|\varphi_i^{\text{QP}}\rangle = \varepsilon_i^{\text{QP}}|\varphi_i^{\text{QP}}\rangle \quad (2)$$

where  $|\varphi_i^{\text{QP}}\rangle$  represents the quasiparticle wave functions. The operator  $\hat{\Sigma}(\cdot)$  is the self-energy operator, which describes the exchange–correlation many-body effects. This operator can be expressed within the GW approximation as a convolution of one-particle Green's function  $G(\mathbf{r}, \mathbf{r}', \omega)$  with the screened Coulomb interaction

$$W(\mathbf{r}, \mathbf{r}', \omega) = \varepsilon^{-1}(\mathbf{r}, \mathbf{r}', \omega)v_{\text{c}}(\mathbf{r}, \mathbf{r}') = \frac{e^{-1}(\mathbf{r}, \mathbf{r}', \omega)}{|\mathbf{r} - \mathbf{r}'|} \quad (3)$$

where  $\varepsilon^{-1}$  is the inverse dielectric function calculated in the random-phase approximation.<sup>36</sup> The self-energy operator can be explicitly written as

$$\Sigma(\mathbf{r}, \mathbf{r}', \omega) = \frac{i}{2\pi} \int d\omega' G(\mathbf{r}, \mathbf{r}', \omega + \omega')W(\mathbf{r}, \mathbf{r}', \omega) \quad (4)$$

Several techniques can be used to perform the frequency integration, starting from separating the self-energy  $\Sigma = iGW$  into its bare exchange part  $\Sigma_{\text{x}} = iGv_{\text{c}}$  and its correlation part  $\Sigma_{\text{c}} = iG\tilde{W}$ , where  $\tilde{W} = W - v_{\text{c}}$ . With this, the integral can be evaluated fully analytically by calculating the reducible polarizability in terms of an eigenvalue decomposition of the random-phase approximation (RPA) Hamiltonian. See, for example, refs 37–383940 for details. While this fully analytical approach (FAA) is analytically exact, it is not feasible for large systems due to the scaling of the diagonalization of the RPA Hamiltonian. Instead, the frequency integration can be approximated within a generalized plasmon-pole model (PPM).<sup>41,42</sup>

One can now by expanding quasiparticle wave functions in terms of Kohn–Sham wave functions transform eq 2 into

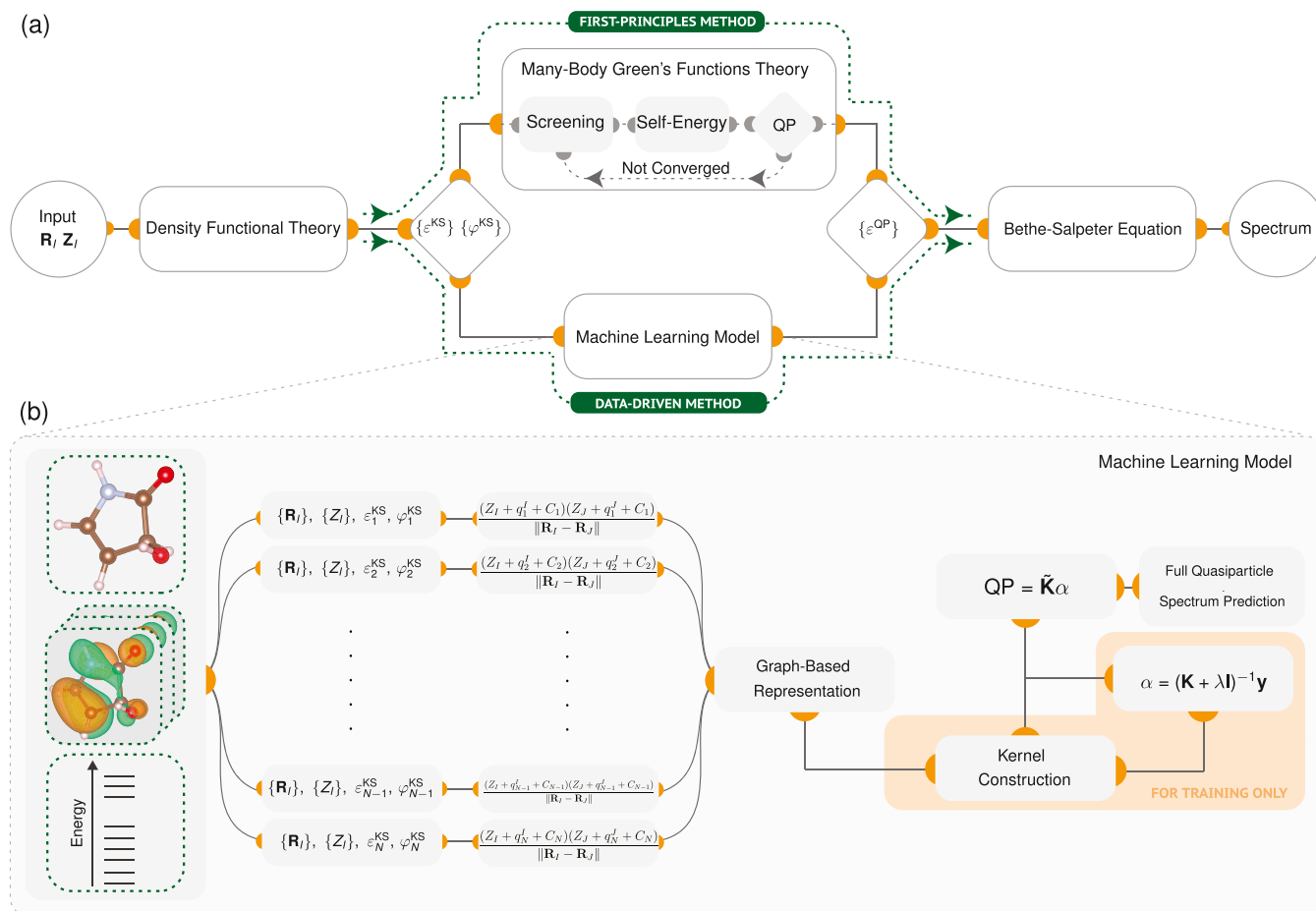
$$H_{ij}^{\text{QP}}(E) = \varepsilon_i^{\text{KS}}\delta_{ij} + \langle \varphi_i^{\text{KS}} | \hat{\Sigma}(E) - \hat{V}_{\text{xc}} | \varphi_j^{\text{KS}} \rangle \quad (5)$$

Further approximating  $|\varphi_i^{\text{QP}}\rangle = |\varphi_i^{\text{KS}}\rangle$ , we can obtain the quasiparticle energies perturbatively as the diagonals of eq 5

$$\begin{aligned} \varepsilon_i^{\text{QP}} &= \varepsilon_i^{\text{KS}} + \Delta\varepsilon_i^{\text{GW}} \\ &= \varepsilon_i^{\text{KS}} + \langle \varphi_i^{\text{KS}} | \hat{\Sigma}(\varepsilon_i^{\text{QP}}) - \hat{V}_{\text{xc}} | \varphi_i^{\text{KS}} \rangle \end{aligned} \quad (6)$$

As  $\Delta\varepsilon_i^{\text{GW}}$  itself depends on  $\varepsilon_i^{\text{QP}}$ , the abovementioned data constitutes a fixed-point problem. In addition, the  $\varepsilon_i^{\text{QP}}$  enters both the energy-dependent microscopic dielectric function determined within the RPA and in Green's function in the expression for the self-energy, eq 4. Hence, in the so-called evGW approach, quasiparticle energies are used to update  $\Sigma$  until self-consistency is reached in combination with eq 6.

Optical excitations of the system lead to the formation of coupled electron–hole pairs, which can be described using a product basis of QP wave functions, that is



**Figure 1.** Schematic overview of first-principles and data-driven routes for the calculation of GW quasiparticle energies and their embedding into a BSE@DFT-GW/ML workflow. (a) Atomic charges and atomic coordinates are used as input for DFT calculations. DFT generates Kohn–Sham energies and wave functions, which are being used as input for either quasiparticle calculations in the first-principles route or  $\Delta$ -ML model in the data-driven route. The output of either route is then used as input for BSE calculations to output the excitation energy spectra. The machine learning block consists of two operations. (b) Details of the  $\Delta$ MLQP model with OSRs. First, the DFT output together with geometric information is transformed into a molecular representation. Second, the resulting vector is used to build up a kernel matrix to predict  $\Delta$ QP energies, which are then added to KS energies and used as output.

$$\chi_S(\mathbf{r}_e, \mathbf{r}_h) = \sum_v^{\text{occ}} \sum_c^{\text{unocc}} \sum_{\sigma\sigma'} A_{vc,\sigma\sigma'}^S \varphi_{c,\sigma'}(\mathbf{r}_e) \varphi_{v,\sigma}^*(\mathbf{r}_h) + B_{vc,\sigma\sigma'}^S \varphi_{v,\sigma'}(\mathbf{r}_e) \varphi_{c,\sigma}^*(\mathbf{r}_h) \quad (7)$$

where  $\mathbf{r}_e$  ( $\mathbf{r}_h$ ) is for the electron (hole) coordinate, and we drop the label QP for clarity. The expansion coefficients  $A_{vc,\sigma\sigma'}$  ( $B_{vc,\sigma\sigma'}$ ) of the excited-state wave function in terms of resonant (anti-resonant) transitions between QP-occupied (occ.) states  $v$  and unoccupied (unocc.)  $c$  with spin  $\sigma$  and  $\sigma'$ , respectively, can be obtained as solutions of the BSE in the form of an effective two-particle Hamiltonian problem

$$\begin{pmatrix} \mathbf{H}^{\text{res}} & \mathbf{M} \\ -\mathbf{M} & -\mathbf{H}^{\text{res}} \end{pmatrix} \begin{pmatrix} \mathbf{A}^S \\ \mathbf{B}^S \end{pmatrix} = \Omega_S \begin{pmatrix} \mathbf{A}^S \\ \mathbf{B}^S \end{pmatrix} \quad (8)$$

In cases with negligible spin–orbit coupling, it can be shown that this Hamiltonian has a block structure in terms of the spin combination of the electron and hole states<sup>43</sup> and can therefore be decomposed into two independent Hamiltonians for singlet and triplet excitations, respectively. This allows us to drop the explicit spin variables, and the matrix elements  $\mathbf{H}^{\text{res}}$  and  $\mathbf{M}$  are determined as

$$H_{vc,v'c'}^{\text{res}} = D_{vc,v'c'} + \kappa M_{vc,v'c'}^x + M_{vc,v'c'}^d \quad (9)$$

$$M_{cv,v'c'} = \kappa M_{cv,v'c'}^x + M_{cv,v'c'}^d \quad (10)$$

where  $\kappa = 2$  (0) for spin singlet (triplet) excitations, and

$$D_{vc,v'c'} = (\varepsilon_c - \varepsilon_v) \delta_{vv'} \delta_{cc'} \quad (11)$$

$$M_{vc,v'c'}^d = - \int d^3\mathbf{r}_e d^3\mathbf{r}_h \varphi_c^*(\mathbf{r}_e) \varphi_{c'}(\mathbf{r}_e) W(\mathbf{r}_e, \mathbf{r}_h, \omega = 0) \times \varphi_v(\mathbf{r}_h) \varphi_{v'}^*(\mathbf{r}_h) \quad (12)$$

$$M_{vc,v'c'}^x = \int d^3\mathbf{r}_e d^3\mathbf{r}_h \varphi_c^*(\mathbf{r}_e) \varphi_v(\mathbf{r}_e) v_c(\mathbf{r}_e, \mathbf{r}_h) \times \varphi_{c'}(\mathbf{r}_h) \varphi_{v'}^*(\mathbf{r}_h) \quad (13)$$

In eq 11, the term  $D$  arises from free interlevel transition between occupied and empty quasiparticle states, and the *direct interaction*  $M^d$  (eq 12) is responsible for the binding of the electron–hole pair and is based on the attractive but screened interaction  $W$  (in the static approximation  $\omega = 0$ ) between

electrons and holes. The repulsive *exchange interaction*  $M^x$  (eq 13) is responsible for the singlet–triplet splitting.

**2.2. Kernel Ridge Regression.** The application of KRR models to predict molecular quantum properties has been very successful over recent years.<sup>8,14,17,23,44</sup> The main idea relies on constructing a kernel matrix with a kernel function  $k$  that can quantitatively measure similarity between molecular representations  $\mathbf{x}_i$  and  $\mathbf{x}_j$ , which are vector representations that encode the molecular physics.<sup>6,24,27</sup> The Laplacian kernel function, for example, is described as

$$k(\mathbf{x}_i, \mathbf{x}_j) = \exp\left(-\frac{\|\mathbf{x}_i - \mathbf{x}_j\|_1}{\sigma}\right) \quad (14)$$

In context of QM, the goal of KRR is to map an input molecular representation  $\mathbf{x}$  to a target quantum property  $p$ . Such a mapping is given by

$$p(\mathbf{x}) = \sum_{n=1}^N \alpha_n k(\mathbf{x}, \mathbf{x}_n) \quad (15)$$

where  $\alpha_n$  stands for the  $n$ th regression coefficient, with  $\mathbf{x}_n$  being the  $n$ th training sample. The learning process within the KRR framework corresponds to obtaining the regression coefficient vector  $\alpha$  for a given reference property vector  $\mathbf{p}^{\text{train}}$ , a kernel matrix  $\mathbf{K}$ , and a regularization coefficient  $\lambda$  by evaluating

$$\alpha = (\mathbf{K} + \lambda \mathbf{I})^{-1} \mathbf{p}^{\text{train}} \quad (16)$$

where  $\mathbf{I}$  is an identity matrix. Additionally, the so-called hyperparameters  $\lambda$  and  $\sigma$  are optimized using the mean absolute error (MAE) metric, where all optimizations are performed with fivefold cross-validation.

**2.3.  $\Delta$ -Machine Learning.** The typical workflow of a supervised machine learning-based task is to map an input vector, for example, a vectorized form of a representation of a molecular geometry, to a target property, such as an atomization or excitation energy, calculated with a high-accuracy method. In some cases, however, it is computationally infeasible to obtain such high-accuracy data for the large number of training/testing inputs required to achieve satisfactory learning results. As a consequence, such *direct* ML approaches can be limited to low-accuracy targets. To circumvent the problem of requiring large high-accuracy data sets,  $\Delta$ -ML aims to predict the highly accurate target property at the same cost of the computationally cheaper methods, which is often referred to as the baseline property.<sup>45,46</sup> This approach is typically more data-efficient than direct ML, since the computationally expensive high-accuracy simulations are needed only for a considerably smaller subset to obtain a certain predictive power.<sup>23,44</sup>

The accurate target property is labeled as  $p^t$  and is obtained by

$$p^t(\mathbf{x}) = p^b(\mathbf{x}) + \Delta_i^t(\mathbf{x}) \quad (17)$$

$$= p^b(\mathbf{x}) + \sum_{n=1}^N \alpha_n k(\mathbf{x}, \mathbf{x}_n) \quad (18)$$

with  $p^b$  being the baseline property and  $\alpha_n$  being the  $n$ th regression coefficient of a KRR model that is trained to predict the difference between the target and baseline property, that is,  $p^t - p^b$ . The  $\Delta$ -ML model has been used in various applications and has shown to be powerful in not only saving computational time but also achieving much higher precision

compared to that of traditional machine learning approaches.<sup>23,46</sup>

### 3. RESULTS

We illustrate the applicability of the orbital-sensitive  $\Delta$  machine learning model for predicting full quasiparticle spectra by reporting the predictive performance on QM8 molecules and water monomers/dimers for molecular Cartesian coordinate-based representations. Subsequently, to obtain excitation energies of acetone and acetone in water, the proposed approach is used as a surrogate model to evaluate the BSE. The results are then used to study solvatochromic shifts and benchmark them against experimental data.

**3.1. Orbital-Sensitive Descriptors.** Figure 1a shows a schematic overview of the  $\Delta$ MLQP approach as a surrogate model to predict quasiparticle energies, thereby bypassing the computationally expensive first-principles GWstep. It is based on the idea of learning the nonlinear transformations from Kohn–Sham energies to quasiparticle energies, motivated by the fact that for all orbitals  $i$

$$\epsilon_i^{\text{QP}} = \epsilon_i^{\text{KS}} + \Delta\epsilon_i \quad (19)$$

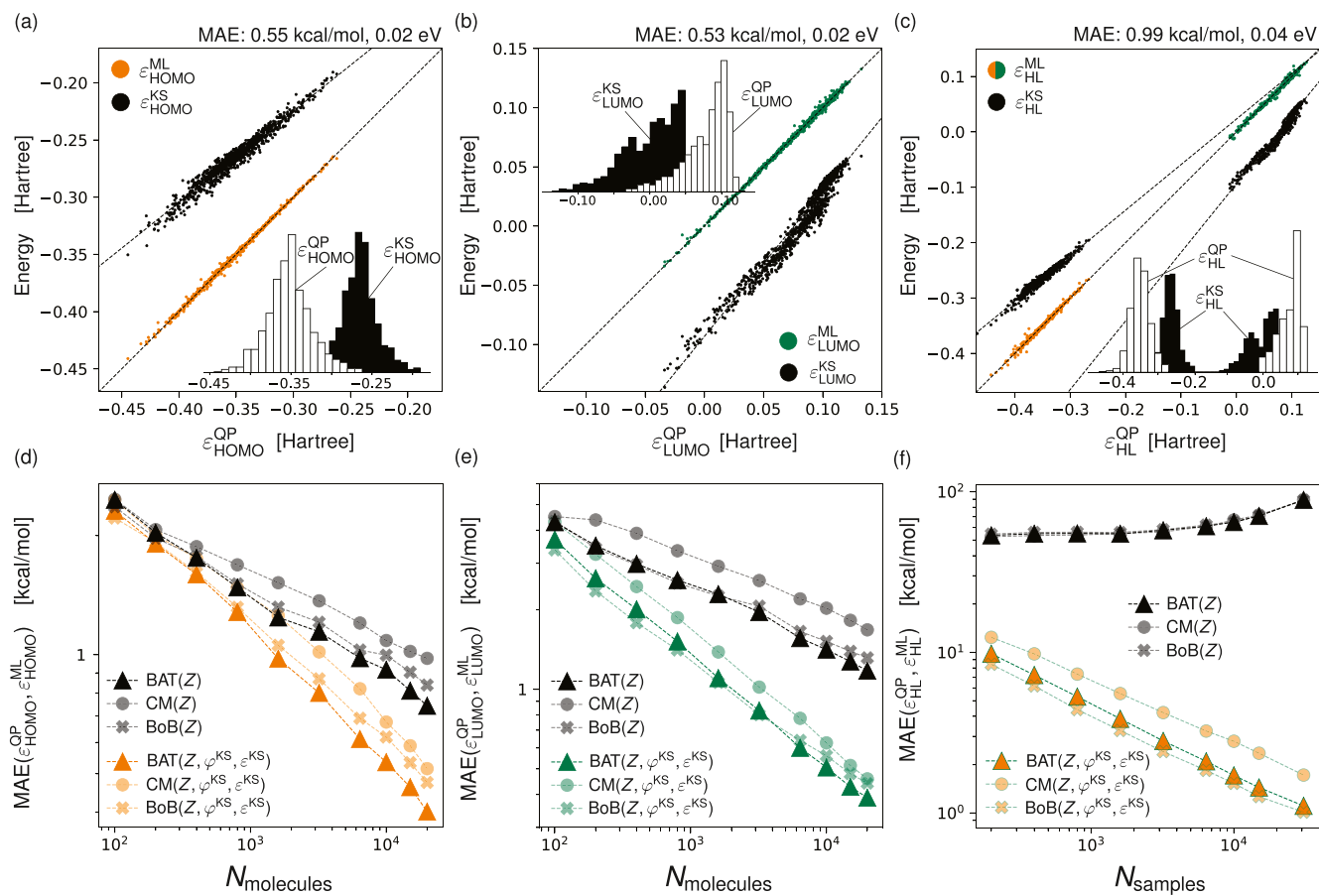
corresponds directly to the form of eq 17. This allows us to identify  $\epsilon_i^{\text{KS}}$  with the baseline property  $p^b(\mathbf{x})$  and to learn the quasiparticle corrections  $\Delta\epsilon_i = \langle \varphi_i^{\text{KS}} | \hat{\Sigma}(\epsilon_i^{\text{QP}}) - \hat{V}_{\text{xc}} | \varphi_i^{\text{KS}} \rangle$  as in eq 6.

To be able to replace all orbital-dependent  $\Delta\epsilon$  by a single Laplacian kernel-based (or any other) machine learning model requires a representation that incorporates orbital information. Traditional graph-based representations, such as the CM and BoB,<sup>2,9,24</sup> which solely rely on the sets of atomic positions  $\{\mathbf{R}\}$  and nuclear charges  $\{Z\}$ , lack such information and have therefore no injectivity for predicting full spectra of molecules.

We propose an extension to the molecular Cartesian coordinate-based representation that includes information about the Kohn–Sham orbital energies and wave functions and will refer to the extended version as OSR  $n$ -OSR, where  $n$  stands for the number of included orbitals. To this end, we map the  $\{\varphi_k^{\text{KS}}(\mathbf{r})\}$  to a set effective orbital-dependent atomic charges  $\{q_k\}$  and add those with its (rescaled) Kohn–Sham energy  $C_k = \zeta \epsilon_k^{\text{KS}}$ , where  $[\zeta] = e/\text{Hartree}$ , to the nuclear charges. For example, the CM representation for a molecule at the electronic state  $k$  can be extended as

$$\text{CM}_k = \begin{cases} \frac{(Z_I + q_k^I + C_k)(Z_J + q_k^J + C_k)}{\|\mathbf{R}_I - \mathbf{R}_J\|}, & \text{for } I \neq J \\ \frac{1}{2}(Z_I + q_k^I + C_k)^{2.4} & \text{for } I = J \end{cases} \quad (20)$$

A modification as in eq 20 allows us to introduce the missing injectivity for multistate predictions within a single model, as indicated in Figure 1b. It is a particularly attractive choice, as all ingredients for this modification are readily available from the DFT baseline calculations, and its very simple, physically interpretable form is easy to implement. The same idea of incorporating orbital sensitivity can be applied to all  $\mathbf{R}$ - and  $Z$ -dependent representations, such as BoB or BAT. In the following, we will evaluate the abovementioned extension of molecular Cartesian coordinate-based representations and their dependence on the choice of different methods to obtain  $\{q_k\}$ , in the use of  $\Delta$ MLQP.



**Figure 2.** Correlation plots and learning curves for  $\Delta$ -machine learning HOMO, LUMO, and HOMO–LUMO energies of QM8 molecules. (a,b) Correlation plots for single-orbital energy level predictions with 1-OSR models, where KS/QP HOMO and LUMO energies are visualized in black with  $\Delta$ -ML/QP HOMO and LUMO energies in orange and green, respectively. (c) Correlation plot for the simultaneous prediction of HOMO and LUMO energies in a 2-OSR model, where KS/QP HOMO/LUMO energies are represented in black with  $\Delta$ -ML in orange/green. The inset in each correlation plot shows the histogram of QP (white) and KS (black) energies. (d–f) HOMO, LUMO, and HOMO–LUMO  $\Delta$ -ML learning curves for QP predictions with various orbital-independent (black) and orbital-dependent representations (orange, green, and orange/green).

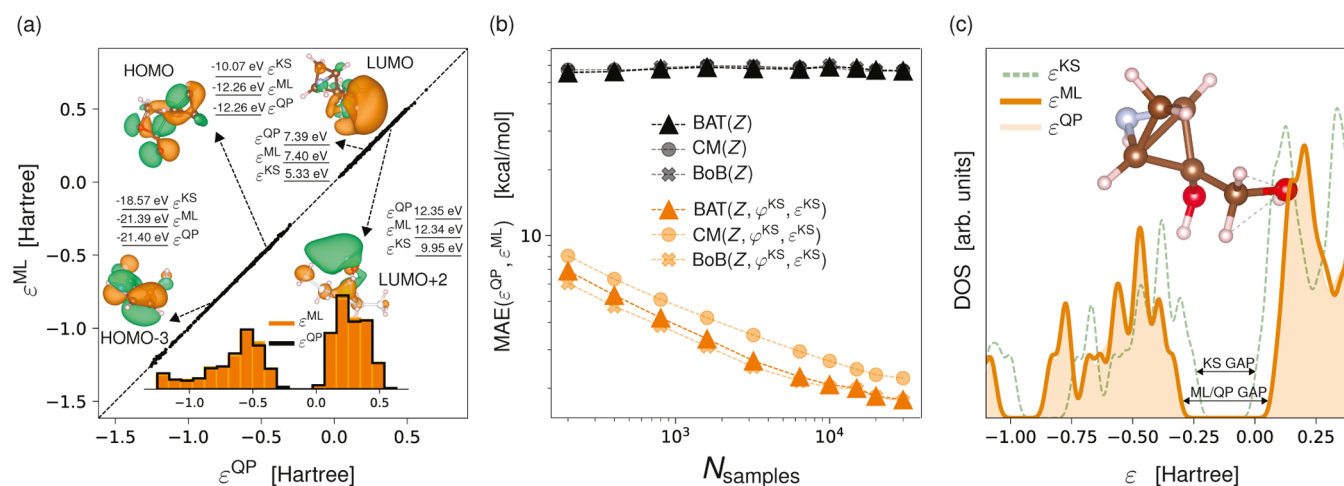
**3.2. QM8 Molecules.** We use molecular geometries from the QM8 data set, which contains more than 20,000 synthetically feasible small organic molecules with up to eight CONF atoms.<sup>22,44</sup> All quantum mechanical calculations have been performed with the VOTCA-XTP package.<sup>39,47</sup> For each molecule, we first perform DFT ground-state calculations with the PBE0 hybrid functional,<sup>48</sup> the def2-TZVP basis set,<sup>49</sup> and an optimized auxiliary basis for the resolution-of-identity techniques.<sup>50</sup> Orbital-dependent atomic charges are determined from a GDMA<sup>29</sup> or from Mulliken populations.<sup>28</sup> Eigenvalue self-consistent (evGW) quasiparticle energies are then determined for the lowest  $2N_{\text{occ}}$  states (excluding the core levels), where  $N_{\text{occ}}$  is the number of occupied levels. All orbitals are included in the RPA step, and not explicitly corrected higher levels are scissor-shifted according to the highest absolute quasiparticle correction among the explicitly corrected unoccupied orbitals. The frequency integration in eq 4 is performed using the FAA.

In Figure 2, we report the performance measures, such as correlations and learning curves, for predicting individual highest occupied molecular orbital (HOMO) energies (Figure 2a,d) and individual lowest unoccupied molecular orbital (LUMO) energies (Figure 2b,e) and predicting simultaneously both HOMO and LUMO energies (Figure 2c,f) of QM8 molecules with various CM, BoB, and BAT representations

and extensions based on GDMA orbital charges. In all cases, here and in the following sections, we used  $\sigma = 800$  and  $\lambda = 10^{-8}$  in the KRR models, and  $\zeta = 1e/\text{Hartree}$ . The hyperparameters were optimized for the QM8 data set with all considered representations, and for the later use cases, it was checked that variations of this optimum have a negligible effect on the predictions.

The correlation and distribution plots display the expected nonlinear shift between KS and QP energies. As expected, the HOMO (LUMO)  $\epsilon^{\text{KS}}$  is consistently above (below) the corresponding  $\epsilon^{\text{QP}}$ . The 1-OSR  $\Delta$ -ML models as in Figure 2a,b are constructed using frontier orbitals of 18,000 molecules as the training set and 2000 samples as the testing set and are based on the extended BAT representations. They transform KS energies to QP energies with a MAE of 0.02 eV. The corresponding learning curves in Figure 2d,e show systematic decay in error with increasing number of training samples  $N_{\text{samples}}$  (here, each sample corresponds to one molecule) for both standard and extended representations. However, it is also clearly visible that even for a 1-OSR model, the inclusion of information about the KS orbital energies and wave functions via the partial charges systematically improves the MAE at fixed training set size in all cases.

We proceed with discussing the first 2-OSR  $\Delta$ MLQP model, trained on a mixed set of 30,000 HOMOs and LUMOs, with



**Figure 3.** Multi-OSR  $\Delta$ -ML of full quasiparticle spectra of QM8 molecules with a single model. (a) Correlation plot of  $\Delta$ -ML/QP energies, where the arrows are pointing at randomly chosen molecules. The inset shows the histogram of ML (orange) and QP (black) energies. (b) Learning curves for QP energy predictions with various orbital-independent (black) and orbital-dependent representations (orange). (c) DOS of a randomly chosen QM8 molecule, where the shaded area (light orange) represents the QP energies, while the green and orange lines describe the KS and ML energies, respectively.

testing performed on 5000. The results of a model simultaneously predicting HOMO and LUMO quasiparticle energies are shown in Figure 2c. Using orbital-sensitive BAT, we note a combined MAE of 0.04 eV. The associated learning curves in Figure 2f show clear improvements of the MAE with increasing number of samples. Note that the training set used to predict multiple states simultaneously contains for each molecule multiple states, and the number of samples does not therefore correspond to the number of molecules. Additionally, unmodified representations fail in predicting both targets at the same time, as expected. Overall, the OSRs are more data-efficient and allow double-state predictions.

Based on the promising performance of the 2-OSR  $\Delta$ -ML model, we apply the same method to predict the full quasiparticle spectra of QM8 molecules in multi-OSR  $\Delta$ MLQP. To do this, we first shuffle the data set comprising all considered quasiparticle levels of all molecules and then select a random subset with 30,000 samples as the training set and 5000 samples as the testing set. In Figure 3a, the ML-QP correlation plot is shown based on application to 1000 out of sample orbitals. Good agreement of the distributions can be observed over an energy range covering around 2 Hartree, where a MAE of 0.06 eV provides a good example of the predictive capabilities of the multi-OSR  $\Delta$ -ML model.

Figure 3b shows the learning curves for standard and OSR. While OSRs allow us to predict the entire QP spectra of QM8 molecules, representations without orbital information fail to learn, and hence, we cannot report systematic improvement with increasing number of training samples. How this translates to a more practical example is shown in Figure 3c. A density of states (DOS) plot of a randomly chosen QM8 molecule (ID 014520 in the QM8 data set) based on KS, QP, and ML is shown. The difference between the KS HOMO–LUMO gap and the QP HOMO–LUMO gap equates to approximately 4 eV, while the difference between the full QP spectra and ML spectra is 0.07 eV (MAE of all predicted levels). The DOS (with a Gaussian broadening of 0.022 Hartree) based on  $\Delta$ MLQP (solid line) is practically indistinguishable from the one based on explicit QP energies (shaded area).

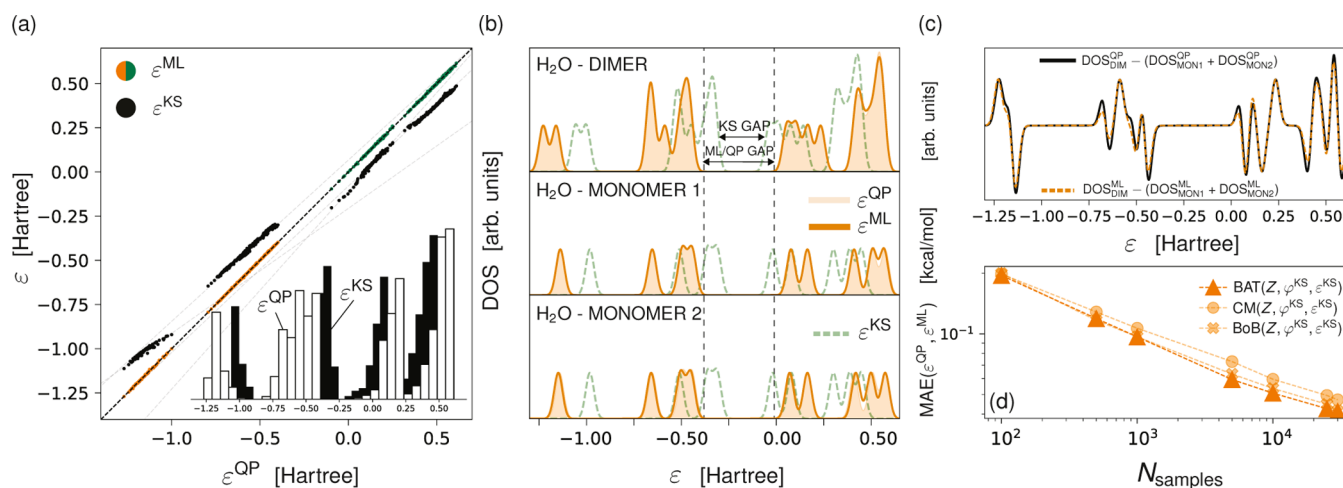
Finally, we compare in Table 1 the MAE of predictions of quasiparticle energies obtained with generic and different

**Table 1.** MAE (in eV) of Predicting *ev*GW Quasiparticle Energies by KRR Using Standard CM, BoB, and BAT Representations and Our Orbital-Sensitive Extension Based on GDMA and Mulliken Orbital Charges<sup>a</sup>

	HOMO	LUMO	2-OSR	multi-OSR
CM	0.043	0.073		
BoB	0.036	0.057		
BAT	0.032	0.051		
CM-GDMA	0.022	0.020	0.046	0.104
BoB-GDMA	0.021	0.019	<b>0.029</b>	<b>0.083</b>
BAT-GDMA	<b>0.017</b>	<b>0.017</b>	0.034	0.089
CM-Mulliken	0.031	0.038	0.109	0.155
BoB-Mulliken	0.026	0.033	0.070	0.108
BAT-Mulliken	0.024	0.030	0.072	0.107

<sup>a</sup>The best-performing models are marked in bold.

OSRs. As also apparent from Figure 2d,e, the addition of orbital information into the representation reduces the MAE even for the single-orbital models by up to 50%. Differences between CM, BoB, and BAT are very small in these cases. Regarding the use of different techniques to obtain effective orbital charges, we note that GDMA yields  $\sim$ 0.010 eV lower MAEs than Mulliken populations, and the overall lowest MAE is obtained for the BAT-GDMA combination. Moving to the 2-OSR- and multi-OSR models, BoB-GDMA performs slightly better than BAT-GDMA and CM-GDMA. In the 2-OSR model comprising the HOMO and LUMO, using Mulliken charges leads to doubling of the MAE, albeit still lower than 0.1 eV. For multi-OSR, the MAEs are expectedly a bit higher due to the high dimensionality of the data, but the differences between the use of GDMA and Mulliken charges are relatively smaller, in particular for BoB and BAT representations. While the Mulliken population-based OSRs appear to yield a slightly higher MAE as compared to the GDMA-based ones, it should be stressed that the latter come with a higher computational cost, noticeable in particular for larger systems (see also



**Figure 4.** Multi-OSR  $\Delta$ -ML of full quasiparticle spectra of water monomers and dimers with a single model. (a) Correlation plot for the simultaneous prediction of all orbital energies, where KS/QP HOMO–LUMO energies are represented in black with  $\Delta$ -ML in orange for the levels below the HOMO–LUMO gap and green for the levels above the gap. The inset shows the histogram of QP (white) and KS (black) energies. (b) DOS plot for a randomly chosen water dimer and its monomers, where the shaded area (light orange) represents the QP energies, while the green and orange lines describe the KS and ML energies, respectively. (c) Difference between the dimer DOS and the monomer DOS summed together, where the QP DOS is represented in black with ML in dashed orange. (d) Learning curves for QP energy predictions with various orbital-dependent representations.

Section 3.4) and are not widely available for standard DFT applications.

**3.3. Water Monomers and Dimers.** In the QM8 example, we focused on predicting the full QP spectra of single molecules. In this example, we want to show that a single  $\Delta$ -ML model can predict the full QP spectra of water dimers and monomers simultaneously, that is,  $\Delta$ MLQP is sensitive to intermolecular conformations as well.

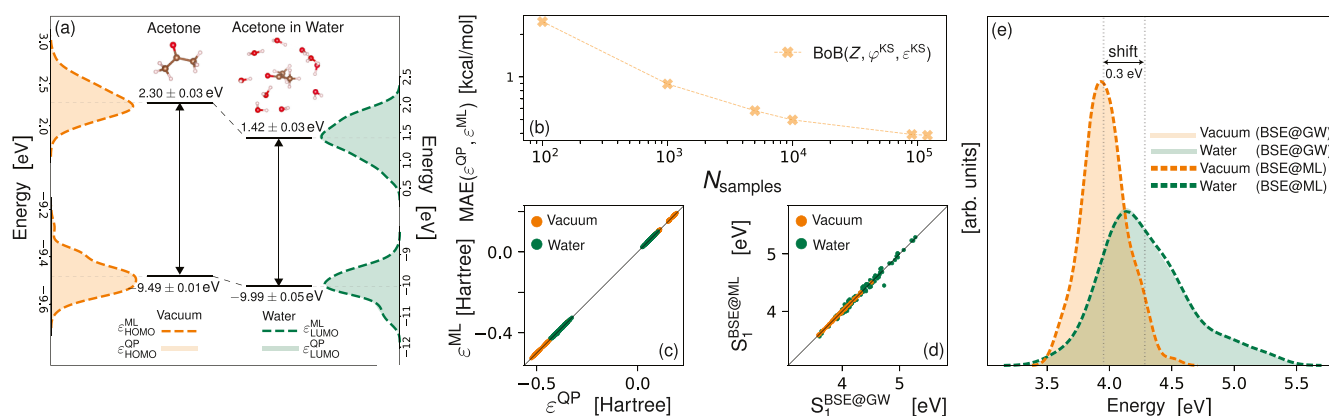
The molecular geometries used in this application originate from the H2O-13 data set<sup>30</sup> that consists of 2000 water dimers with O–O distances less than 4.5 Å obtained from an MD simulation. All GW properties used in the training and testing sets were calculated as mentioned for the QM8 data in the previous section. To build a multi-OSR  $\Delta$ MLQP model, we first not only consider the 2000 dimer structures but also extract 4000 monomers from them. In water dimers (monomers), 18 quasiparticle states are taken into account, so the orbital data set we have used contains 52,000 samples in total. From this set, 30,000 randomly selected samples are used for training and 5k for testing.

Figure 4a shows the QP–KS correlation (black) and QP–ML correlation (orange/green). The multi-OSR  $\Delta$ MLQP approach based on BAT is able to transform the Kohn–Sham energies to quasiparticle energies with great accuracy (0.03 eV MAE). In Figure 4b, we plot the DOS of a randomly chosen water dimer and its constituent monomers broadened by 0.022 Hartree. On the shown scale, the small differences among the two monomers are hard to distinguish, independent of the method. More importantly, it is apparent that  $\Delta$ MLQP captures the respective openings of the HOMO–LUMO gaps and energy-dependent quasiparticle corrections and the effects of intermolecular interactions in the dimer conformation. This point is emphasized by the analysis of the difference between the actual dimer DOS and a simple superposition of the two monomer DOS as in the top panel of Figure 4c, evaluated based on the explicit QP energies (solid black) and the one predicted by  $\Delta$ MLQP (dashed). These differences reveal the shifts of the coupled dimer energy levels due to

intermolecular interactions, and the effects are captured by  $\Delta$ MLQP at very good accuracy not only near the gap but also for, for example, the deep O 2s levels. The lower panel in Figure 4c shows monotonous decay of the prediction errors as a function of training set size, where the orbital-sensitive BoB and BAT perform slightly better than the orbital-sensitive CM. Again, we see from the correlation plots, DOS, and the learning curves that the full QP spectra of water dimers and monomers can be accurately reproduced with just a single delta machine learning model.

**3.4. Acetone in Water.** We now consider an example of even more complex molecular clusters: aqueous acetone. The increased complexity stems from combining two different molecular species and considering more molecules in the clusters, leading to a very high-dimensional problem as the number of states and conformations increases dramatically. Specifically, the choice for aqueous acetone is motivated by the fact that it is known to exhibit a solvatochromic shift of the lowest coupled electron–hole excitation energy of  $\sim 0.2$  eV,<sup>51–53</sup> a combined effect of similar shifts to the individual quasiparticle energies and modified screening of the electron–hole interaction in water. From the perspective of our  $\Delta$ MLQP model, this poses the additional question of whether its predictions are accurate enough in such a case to embed them into the calculation of the electron–hole excitation energies via the BSE (BSE@ML vs BSE@GW), as noted in the workflow scheme in Figure 1a.

To answer this question, we first generate structural data by performing classical molecular dynamics simulations of single acetone in 219 water molecules using an OPLS-AA-type force field for acetone, automatically generated using LigParGen,<sup>54</sup> and the TIP3P model for water.<sup>55</sup> Geometric mixing rules for Lennard-Jones diameters and energies were used for atoms of different species.<sup>56</sup> Nonbonded interactions between atom pairs within a molecule separated by one or two bonds were excluded. The interaction was reduced by a factor of 1/2 for atoms separated by three bonds and more. Simulations were run using GROMACS version 2019.6.<sup>57</sup> A cutoff of 0.9 nm was



**Figure 5.** Multi-OSR  $\Delta$ -ML of full quasiparticle spectra of acetone and acetone in water with a single model and application to calculating  $n \rightarrow \pi^*$  excitation energies with BSE. (a) The black bars are a visual representation of the HOMO–LUMO gap in acetone in vacuum (orange) and acetone in water (green) with ML predictions. The shaded areas represent the HOMO and LUMO distributions resulting from explicit evGW calculations, while the ones resulting from the 2-OSR  $\Delta$ MLQP model are shown as dashed lines. (b) Prediction errors as a function of training set size for the 8-OSR-BoB model to be used in BSE@ML calculations. (c) QP-ML quasiparticle energy correlation for 8-OSR-BoB. (d) Correlation between  $S_1$  ( $n \rightarrow \pi^*$ ) energies from BSE@GW and BSE@ML. (e)  $S_1$  energy distributions of a single acetone (orange) and acetone in water (green) from BSE@GW and BSE@ML.

employed for the real space part of electrostatics and Lennard-Jones interactions. The long-range electrostatics were calculated using particle-mesh Ewald<sup>58</sup> with the reciprocal space interactions evaluated on a 0.18 nm grid with cubic interpolation of order 4. An initial configuration was prepared in a cubic box of 2 nm size and energy-minimized using the steepest descent algorithm, followed by a 6 ns simulation in a constant-particle number, volume, and temperature (NVT) ensemble. Temperature was kept constant at 300 K using the stochastic velocity-rescaling thermostat<sup>59</sup> with a time constant of 0.5 ps, and the velocity-Verlet algorithm<sup>60</sup> was employed to integrate the equations of motions with a 1 fs time step. Simulations were then continued for 200 ns in a constant-particle number, pressure, and temperature ( $NpT$ ) ensemble at 300 K and 1 bar controlled using the Berendsen<sup>61</sup> barostat with a coupling time constant of 2.0 ps. From the last 100 ns of this run, 10,000 snapshots at a time interval of 10 ps are extracted, and clusters containing acetone and the 10 water molecules closest to it are selected for the GW-BSE calculations. With this choice, we ascertain that the first solvation shell is included in the cluster, contributing the strongest to the expected solvatochromic shifts. Note that we are not strictly targeting quantitative accuracy of the actual excitation energies compared to the experiment with this study, but we strive to demonstrate the internal consistency between BSE@GW and BSE@ML in our multi-OSR  $\Delta$ MLQP approach.

GW-BSE calculations are performed on the selected clusters without and with the water molecules included, following the same protocol as before, with the exception of the treatment of the frequency dependence in eq 4, for which we employ here a two-parameter-generalized PPM.<sup>41,42,62</sup> Explicit quasiparticle corrections are determined for the four HOMOs and four LUMOs, while the full single-particle spectrum is included in the formation of the product basis for the BSE.

We first construct as a proof of concept a 2-OSR-BoB model including the HOMO and LUMO in the absence and presence of a water solvation shell. From the total of 80,000 samples, we select 5000 for training and 1000 for testing, ensuring that the sets have an equal amount of data for both the HOMO and

LUMO, with and without water, respectively. The energy distributions as obtained from the explicit calculations of  $\epsilon^{\text{QP}}$  in vacuum (in water), calculated on 400 (100 of each case) out of sample data points, are shown as filled orange (green) areas in Figure 5a. Indicated are also the means and their error, showing a distinct lowering of 0.50 and 0.92 eV for the HOMO and LUMO, respectively, in the presence of water, thereby decreasing the HOMO–LUMO gap by 0.42 eV. Comparing the respective distributions obtained from the 2-OSR  $\Delta$ MLQP model shown as dashed lines in Figure 5a, hardly, any differences can be observed.

As mentioned above, the determination of electron–hole excitation energies with BSE@GW requires inclusion of eight explicitly corrected orbitals near the gap. For this purpose, we now build an 8-OSR  $\Delta$ MLQP model, for which Figure 5b shows the learning curves. With more than 10k samples, a MAE of lower than 0.5 kcal/mol can be achieved. Figure 5c shows the corresponding correlation between explicitly calculated QP energies and the ones from 8-OSR  $\Delta$ MLQP, for which we have selected from the full data set comprising 320k orbitals 120k for training and 10k for testing. Clearly, the  $\Delta$ MLQP model again provides excellent predictions for the eight different orbital energies in vacuum and solution. In the following step, these predicted quasiparticle energies for the eight explicitly corrected orbitals are used as input for the BSE. The remaining occupied (unoccupied) single-particle energies are scissor-shifted according to the highest absolute quasiparticle correction among the explicitly corrected occupied (unoccupied) orbitals, as in the BSE@GW reference. In Figure 5d, we show the correlation between the determined  $S_1$  energies of the  $n \rightarrow \pi^*$  transition. Two interesting aspects should be noted: the subset of results for BSE@ML for the vacuum structure (orange) appears to agree better than the one for the acetone–water clusters (green), which is not surprising due to the bigger conformational space of the latter. More importantly, however, one can clearly see a systematic shift of the  $S_1$  energies in aqueous solution to higher energies compared to those in vacuo—equally obtained for both BSE@GW and BSE@ML—qualitatively in line with the experimental observations.



Finally, we show in Figure 5e distributions of the  $n \rightarrow \pi^*$  excitation energies of acetone as obtained by the BSE@GW (filled curves) and BSE@ML (dashed lines) approaches in vacuum (orange) and in water (green), respectively. It is evident that both methods predict both a broadening of the distribution upon solvation and a shift to higher energies. Differences of the distributions in BSE@GW and BSE@ML levels are miniscule. The predicted solvatochromic shift of the mean of the distributions amounts to 0.30 eV, as indicated by the dashed lines in Figure 5e. For peak-to-peak, the shift is 0.13 eV.

#### 4. SUMMARY

We have introduced orbital-sensitive augmentations of molecular Cartesian coordinate-based representations in  $\Delta$ -machine learning of full quasiparticle excitation energies in molecules and clusters. The proposed  $\Delta$ MLQP approach is capable of predicting the GW energies of multiple orbitals across multiple molecules and/or intra- and intermolecular conformations in a single KRR-based supervised learning model. We have demonstrated this in application to the QM8 molecular data set and to water monomers and dimers. Furthermore, it has been shown that a single orbital-sensitive  $\Delta$ -ML model for quasiparticle energies can be embedded in multiscale simulation workflows, showcased in the prediction of solvatochromic shifts of excitation energies in aqueous acetone.

#### AUTHOR INFORMATION

##### Corresponding Author

**Björn Baumeier** – Department of Mathematics and Computer Science, Eindhoven University of Technology, 5600MB Eindhoven, The Netherlands; Institute for Complex Molecular Systems, Eindhoven University of Technology, 5600MB Eindhoven, The Netherlands; [orcid.org/0000-0002-6077-0467](https://orcid.org/0000-0002-6077-0467); Email: [b.baumeier@tue.nl](mailto:b.baumeier@tue.nl)

##### Author

**Onur Çaylak** – Department of Mathematics and Computer Science, Eindhoven University of Technology, 5600MB Eindhoven, The Netherlands; Institute for Complex Molecular Systems, Eindhoven University of Technology, 5600MB Eindhoven, The Netherlands

Complete contact information is available at: <https://pubs.acs.org/10.1021/acs.jctc.1c00520>

##### Notes

The authors declare no competing financial interest. The raw data of the GW calculations on the QM8 data set, water monomers and dimers, and aqueous acetone are available as VOTCA-XTP output files at <https://dx.doi.org/10.6084/m9.figshare.14625564>.

#### ACKNOWLEDGMENTS

We acknowledge support by the Innovational Research Incentives Scheme Vidi of the Netherlands Organisation for Scientific Research (NWO) with project number 723.016.002. Partial funding is also provided by the NWO and the Netherlands eScience Center through project number 027.017.G15, within the Joint CSER and eScience program for Energy Research (JCER 2017). We thank Ruben Gerritsen for critical reading of the manuscript.

#### REFERENCES

- (1) von Lilienfeld, O. A. *Many-Electron Approaches in Physics, Chemistry and Mathematics: A Multidisciplinary View*; Bach, V., Delle Site, L., Eds.; Mathematical Physics Studies; Springer International Publishing: Cham, 2014; pp 169–189.
- (2) Huang, B.; Lilienfeld, O. A. V. Communication: Understanding Molecular Representations in Machine Learning: The Role of Uniqueness and Target Similarity. *J. Chem. Phys.* **2016**, *145*, 241730.
- (3) Bartók, A. P.; Kermode, J.; Bernstein, N.; Csányi, G. Machine Learning a General-Purpose Interatomic Potential for Silicon. *Phys. Rev. X* **2018**, *8*, 041048.
- (4) Collins, C. R.; Gordon, G. J.; von Lilienfeld, O. A.; Yaron, D. J. Constant Size Descriptors for Accurate Machine Learning Models of Molecular Properties. *J. Chem. Phys.* **2018**, *148*, 241718.
- (5) Schütt, K. T.; Gastegger, M.; Tkatchenko, A.; Müller, K. R.; Maurer, R. J. Unifying Machine Learning and Quantum Chemistry with a Deep Neural Network for Molecular Wavefunctions. *Nat. Commun.* **2019**, *10*, 5024.
- (6) Çaylak, O.; Yaman, A.; Baumeier, B. Evolutionary Approach to Constructing a Deep Feedforward Neural Network for Prediction of Electronic Coupling Elements in Molecular Materials. *J. Chem. Theory Comput.* **2019**, *15*, 1777–1784.
- (7) Ghosh, K.; Stuke, A.; Todorović, M.; Jørgensen, P. B.; Schmidt, M. N.; Vehtari, A.; Rinke, P. Deep Learning Spectroscopy: Neural Networks for Molecular Excitation Spectra. *Adv. Sci.* **2019**, *6*, 1801367.
- (8) Çaylak, O.; Anatole von Lilienfeld, O.; Baumeier, B. Wasserstein Metric for Improved Quantum Machine Learning with Adjacency Matrix Representations. *Mach. Learn.: Sci. Technol.* **2020**, *1*, 03LT01.
- (9) von Lilienfeld, O. A.; Burke, K. Retrospective on a Decade of Machine Learning for Chemical Discovery. *Nat. Commun.* **2020**, *11*, 4895.
- (10) Wang, A. Y.-T.; Murdock, R. J.; Kauwe, S. K.; Oliynyk, A. O.; Gurlo, A.; Brgoch, J.; Persson, K. A.; Sparks, T. D. Machine Learning for Materials Scientists: An Introductory Guide toward Best Practices. *Chem. Mater.* **2020**, *32*, 4954–4965.
- (11) Song, Z.; Chen, X.; Meng, F.; Cheng, G.; Wang, C.; Sun, Z.; Yin, W.-J. Machine Learning in Materials Design: Algorithm and Application. *Chin. Phys. B* **2020**, *29*, 116103.
- (12) Tkatchenko, A. Machine Learning for Chemical Discovery. *Nat. Commun.* **2020**, *11*, 4125.
- (13) von Lilienfeld, O. A.; Müller, K.-R.; Tkatchenko, A. Exploring Chemical Compound Space with Quantum-Based Machine Learning. *Nat. Rev. Chem.* **2020**, *4*, 347–358.
- (14) Lemm, D.; von Rudorff, G. F.; von Lilienfeld, O. A. Machine learning based energy-free structure predictions of molecules, transition states, and solids. *Nat. Commun.* **2021**, *12*, 4468.
- (15) Westermayr, J.; Marquetand, P. Machine Learning for Electronically Excited States of Molecules. *Chem. Rev.* **2020**, DOI: [10.1021/acs.chemrev.0c00749](https://doi.org/10.1021/acs.chemrev.0c00749).
- (16) Westermayr, J.; Maurer, R. J. Physically Inspired Deep Learning of Molecular Excitations and Photoemission Spectra. *Chem. Sci.* **2021**, DOI: [10.1039/D1SC01542G](https://doi.org/10.1039/D1SC01542G).
- (17) Tirimbó, G.; Çaylak, O.; Baumeier, B. A Kernel-Based Machine Learning Approach to Computing Quasiparticle Energies within Many-Body Green's Functions Theory. Machine Learning for Molecules Workshop NeurIPS. **2020**, arxiv:2012.01787.
- (18) Juan, Y.; Dai, Y.; Yang, Y.; Zhang, J. Accelerating Materials Discovery Using Machine Learning. *J. Mater. Sci. Technol.* **2021**, *79*, 178–190.
- (19) Rauer, C.; Bereau, T. Hydration Free Energies from Kernel-Based Machine Learning: Compound-Database Bias. *J. Chem. Phys.* **2020**, *153*, 014101.
- (20) Dong, S. S.; Govoni, M.; Galli, G. Machine Learning Dielectric Screening for the Simulation of Excited State Properties of Molecules and Materials. *Chem. Sci.* **2021**, *12*, 4970–4980.
- (21) Dral, P. O.; Barbatti, M. Molecular Excited States through a Machine Learning Lens. *Nat. Rev. Chem.* **2021**, *5*, 388–405.

- (22) Ramakrishnan, R.; Dral, P. O.; Rupp, M.; von Lilienfeld, O. A. Quantum Chemistry Structures and Properties of 134 Kilo Molecules. *Sci. Data* **2014**, *1*, 140022.
- (23) Ramakrishnan, R.; Dral, P. O.; Rupp, M.; von Lilienfeld, O. A. Big Data Meets Quantum Chemistry Approximations: The  $\Delta$ -Machine Learning Approach. *J. Chem. Theory Comput.* **2015**, *11*, 2087–2096.
- (24) Hansen, K.; Biegler, F.; Ramakrishnan, R.; Pronobis, W.; von Lilienfeld, O. A.; Müller, K.-R.; Tkatchenko, A. Machine Learning Predictions of Molecular Properties: Accurate Many-Body Potentials and Nonlocality in Chemical Space. *J. Phys. Chem. Lett.* **2015**, *6*, 2326–2331.
- (25) Meftahi, N.; Klymenko, M.; Christofferson, A. J.; Bach, U.; Winkler, D. A.; Russo, S. P. Machine Learning Property Prediction for Organic Photovoltaic Devices. *npj Comput. Mater.* **2020**, *6*, 166.
- (26) Liu, Z.; Lin, L.; Jia, Q.; Cheng, Z.; Jiang, Y.; Guo, Y.; Ma, J. Transferable Multilevel Attention Neural Network for Accurate Prediction of Quantum Chemistry Properties via Multitask Learning. *J. Chem. Inf. Model.* **2021**, *61*, 1066–1082.
- (27) Rupp, M.; Tkatchenko, A.; Müller, K. R.; von Lilienfeld, O. A. Fast and Accurate Modeling of Molecular Atomization Energies with Machine Learning. *Phys. Rev. Lett.* **2012**, *108*, 058301.
- (28) Mulliken, R. S. Electronic Population Analysis on LCAO–MO Molecular Wave Functions. I. *J. Chem. Phys.* **1955**, *23*, 1833–1840.
- (29) Stone, A. J. Distributed Multipole Analysis: Stability for Large Basis Sets. *J. Chem. Theory Comput.* **2005**, *1*, 1128–1132.
- (30) Bartók, A. P.; Gillan, M. J.; Manby, F. R.; Csányi, G. Machine-Learning Approach for One- and Two-Body Corrections to Density Functional Theory: Applications to Molecular and Condensed Water. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, *88*, 054104.
- (31) Kohn, W.; Sham, L. J. Self-Consistent Equations Including Exchange and Correlation Effects. *Phys. Rev.* **1965**, *140*, A1133–A1138.
- (32) Hedin, L. New Method for Calculating the One-Particle Green's Function with Application to the Electron-Gas Problem. *Phys. Rev.* **1965**, *139*, A796–A823.
- (33) Hedin, L.; Lundqvist, S. *Solid State Physics*; Seitz, F., Turnbull, D., Ehrenreich, H., Eds.; Academic Press, 1970; Vol. 23, pp 1–181.
- (34) Aulbur, W. G.; Jönsson, L.; Wilkins, J. W. *Solid State Physics*; Ehrenreich, H., Spaepen, F., Eds.; Academic Press, 2000; Vol. 54, pp 1–218.
- (35) Rohlfing, M. Excited States of Molecules from Green's Function Perturbation Techniques. *Int. J. Quantum Chem.* **2000**, *80*, 807–815.
- (36) Hybertsen, M. S.; Louie, S. G. First-Principles Theory of Quasiparticles: Calculation of Band Gaps in Semiconductors and Insulators. *Phys. Rev. Lett.* **1985**, *55*, 1418–1421.
- (37) Golze, D.; Wilhelm, J.; van Setten, M. J.; Rinke, P. Core-Level Binding Energies from GW: An Efficient Full-Frequency Approach within a Localized Basis. *J. Chem. Theory Comput.* **2018**, *14*, 4856–4869.
- (38) Golze, D.; Dvorak, M.; Rinke, P. The GW Compendium: A Practical Guide to Theoretical Photoemission Spectroscopy. *Front. Chem.* **2019**, *7*, 377.
- (39) Tirimbò, G.; Sundaram, V.; Çaylak, O.; Scharpach, W.; Sijen, J.; Junghans, C.; Brown, J.; Ruiz, F. Z.; Renaud, N.; Wehner, J.; Baumeier, B. Excited-State Electronic Structure of Molecules Using Many-Body Green's Functions: Quasiparticles and Electron–Hole Excitations with VOTCA-XTP. *J. Chem. Phys.* **2020**, *152*, 114103.
- (40) Çaylak, O.; Baumeier, B. Excited-State Geometry Optimization of Small Molecules with Many-Body Green's Functions Theory. *J. Chem. Theory Comput.* **2021**, *17*, 879–888.
- (41) Hybertsen, M. S.; Louie, S. G. Electron Correlation in Semiconductors and Insulators: Band Gaps and Quasiparticle Energies. *Phys. Rev. B* **1986**, *34*, 5390–5413.
- (42) Godby, R. W.; Needs, R. J. Metal-Insulator Transition in Kohn-Sham Theory and Quasiparticle Theory. *Phys. Rev. Lett.* **1989**, *62*, 1169–1172.
- (43) Rohlfing, M.; Louie, S. G. Electron-Hole Excitations and Optical Spectra from First Principles. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2000**, *62*, 4927–4944.
- (44) Ramakrishnan, R.; Hartmann, M.; Tapavicza, E.; Lilienfeld, O. A. V. Electronic Spectra from TDDFT and Machine Learning in Chemical Space. *J. Chem. Phys.* **2015**, *143*, 084111.
- (45) Rupp, M. Machine Learning for Quantum Mechanics in a Nutshell. *Int. J. Quantum Chem.* **2015**, *115*, 1058–1073.
- (46) Zaspel, P.; Huang, B.; Harbrecht, H.; von Lilienfeld, O. A. Boosting Quantum Machine Learning Models with a Multilevel Combination Technique: Pople Diagrams Revisited. *J. Chem. Theory Comput.* **2019**, *15*, 1546–1559.
- (47) Wehner, J.; Brombacher, L.; Brown, J.; Junghans, C.; Çaylak, O.; Khalak, Y.; Madhikar, P.; Tirimbò, G.; Baumeier, B. Electronic Excitations in Complex Molecular Environments: Many-Body Green's Functions Theory in VOTCA-XTP. *J. Chem. Theory Comput.* **2018**, *14*, 6253–6268.
- (48) Adamo, C.; Barone, V. Toward Reliable Density Functional Methods without Adjustable Parameters: The PBE0 Model. *J. Chem. Phys.* **1999**, *110*, 6158–6170.
- (49) Weigend, F.; Ahlrichs, R. Balanced Basis Sets of Split Valence, Triple Zeta Valence and Quadruple Zeta Valence Quality for H to Rn: Design and Assessment of Accuracy. *Phys. Chem. Chem. Phys.* **2005**, *7*, 3297–3305.
- (50) Weigend, F.; Häser, M.; Patzelt, H.; Ahlrichs, R. RI-MP2: Optimized Auxiliary Basis Sets and Demonstration of Efficiency. *Chem. Phys. Lett.* **1998**, *294*, 143–152.
- (51) Bayliss, N. S.; McRae, E. G. Solvent Effects in the Spectra of Acetone, Crotonaldehyde, Nitromethane and Nitrobenzene. *J. Phys. Chem.* **1954**, *58*, 1006–1011.
- (52) Merchán, M.; Roos, B. O.; McDiarmid, R.; Xing, X. A Combined Theoretical and Experimental Determination of the Electronic Spectrum of Acetone. *J. Chem. Phys.* **1996**, *104*, 1791–1804.
- (53) Crescenzi, O.; Pavone, M.; De Angelis, F.; Barone, V. Solvent Effects on the UV ( $n \rightarrow \pi^*$ ) and NMR (13C and 17O) Spectra of Acetone in Aqueous Solution. An Integrated Car-Parrinello and DFT/PCM Approach. *J. Phys. Chem. B* **2005**, *109*, 445–453.
- (54) Dodda, L. S.; Cabeza de Vaca, I.; Tirado-Rives, J.; Jorgensen, W. L. LigParGen Web Server: An Automatic OPLS-AA Parameter Generator for Organic Ligands. *Nucleic Acids Res.* **2017**, *45*, W331–W336.
- (55) Jorgensen, W. L.; Chandrasekhar, J.; Madura, J. D.; Impey, R. W.; Klein, M. L. Comparison of Simple Potential Functions for Simulating Liquid Water. *J. Chem. Phys.* **1983**, *79*, 926–935.
- (56) Jorgensen, W. L.; Maxwell, D. S.; Tirado-Rives, J. Development and Testing of the OPLS All-Atom Force Field on Conformational Energetics and Properties of Organic Liquids. *J. Am. Chem. Soc.* **1996**, *118*, 11225–11236.
- (57) Abraham, M. J.; Murtola, T.; Schulz, R.; Páll, S.; Smith, J. C.; Hess, B.; Lindahl, E. GROMACS: High Performance Molecular Simulations through Multi-Level Parallelism from Laptops to Supercomputers. *SoftwareX* **2015**, *1–2*, 19–25.
- (58) Essmann, U.; Perera, L.; Berkowitz, M. L.; Darden, T.; Lee, H.; Pedersen, L. G. A Smooth Particle Mesh Ewald Method. *J. Chem. Phys.* **1995**, *103*, 8577–8593.
- (59) Bussi, G.; Donadio, D.; Parrinello, M. Canonical Sampling through Velocity Rescaling. *J. Chem. Phys.* **2007**, *126*, 014101.
- (60) Verlet, L. Computer "Experiments" on Classical Fluids. I. Thermodynamical Properties of Lennard-Jones Molecules. *Phys. Rev.* **1967**, *159*, 98–103.
- (61) Berendsen, H. J. C.; Postma, J. P. M.; van Gunsteren, W. F.; DiNola, A.; Haak, J. R. Molecular Dynamics with Coupling to an External Bath. *J. Chem. Phys.* **1984**, *81*, 3684–3690.
- (62) Rohlfing, M.; Krüger, P.; Pollmann, J. Efficient Scheme for GW Quasiparticle Band-Structure Calculations with Applications to Bulk Si and to the Si(001)-(2x1) Surface. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1995**, *52*, 1905–1917.